Appasaheb Ragunathrao Bhaurao Garud Arts, Commerce and Science College, Shendurni.

Pre/Post Test for bridge course
Slow And Advance learner
Academic Year :2015-2016

Subject: Chemistry **Class:** F. Y. B. Sc. **Month:** September (Incl Sun

Roll No	Student Name	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	Total	Sign Of Student
1	BANKAR BHUSHAN SANJEEV	P	P	P	P	P	P	.	.	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	.	.	P	P	P	P		23	
2	BANKAR PRADIP SHANTARAM	.	P	.	P	P	P	P	P	.	P	P	.	P	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		23	
3	BHADANE MADHURI RAJENDRA	P	P	.	P	P	P	P	P	P	P	P	P	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.		27	
4	BHAVSAR JITENDRA PRAMOD	P	P	P	P	.	P	.	.	.	P	P	P	P	P	.	.	P	P	P	.	.	.	P	P	P	P	P	P	P	P		21	
5	BHOI VIKRAM RAMLAL	P	P	P	P	P	P	P	P	P	.	P	P	P	P	P	.	P	P	P	P	P	P	P	P	P	P	P	P	.	P		27	
6	BUNDE NAMRATA VASUDEO	–	–	P	P	P	P	P	P	–	P	P	P	P	P	P	P	–	–	P	P	P	P	P	P	P	P	P	P	P	P		25	
7	CHAUDHARI KALYANI RAJENDRA	.	.	.	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	.	.	P	P	.	.	.	P	P	P	P	P		20	
8	CHAUDHARI NIKITA DINESH	P	P	P	–	P	P	P	P	P	P	P	–	P	–	P	P	P	P	P	P	P	–	–	P	P	P	–	P	P	P		24	
9	CHAUDHARI PUJA AMRUT	P	P	P	P	P	.	P	P	P	P	P	P	P	P	P	.	.	P	P	P	P	.	P	P	P	P	P	.	P	P		25	
10	CHAUDHARI RAHUL NANA	P	P	P	P	P	P	–	P	P	P	P	P	P	P	–	–	P	P	P	P	P	P	–	P	P	P	P	P	P	P		26	
11	CHAUDHARI RUPALI GAJANAN	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		30	
12	CHAVHAN DIPALI RAJARAM	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	P	P	.	.	P	P	P	P	P	P	P	.	P		24	
13	GHONGADE YOGITA ISHWAR	.	.	.	P	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		24	
14	GUJAR NAINA KISHOR	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	P	P		26	
15	GUJAR PRIYANKA RAMESH	.	P	P	P	P	P	P	P	P	P	P	P	.	P	P	P	P	P	P	.	.	.	.	P	P	P	P	P	P	P		24	
16	JADHAV BANDU DEVIDAS	–	–	P	P	P	P	P	P	P	P	–	P	P	P	P	P	P	–	P	P	P	P	P	–	–	P	P	–	P	P		23	
17	JADHAV RITESH KESHAV	P	P	P	P	P	.	.	P	P	P	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		27	
18	JANGALE RAJSHRI SANJAY	.	.	P	P	P	P	P	.	.	P	.	P	P	P	P	P	.	.	P	P	.	.	.	P	P	P	P	P	P	P		20	
19	JOSHI MANASI DILEEP	.	.	.	.	.	P	P	P	P	P	P	.	P	P	P	P	P	P	P	P	.	P	P	P	P	P	P	P	P	P		23	
20	JOSHI RADHA SHESHRAO	.	.	P	P	.	P	P	.	P	P	P	P	.	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P		22	
21	KAKADE SACHIN VASANT																																	
22	KATE PRATIKSHA PRAKASH																																	
23	MAHAJAN MAMATA VASANTRAO	.	.	.	P	P	P	.	.	.	.	P	P	P	.	.	.	.	P	P	P	P	P	.	.	P	P	.	P	P	P		16	
24	MARATHE VIDYA SUKDEV	P	P	P	.	P	P	P	.	.	P	P	P	P	P	P	.	P	P	P	P	P	.	P	P	.	.	P	P	P	P		23	
25	NAIK SHRIKRUSHNA SUDAM	P	P	P	P	.	P	P	P	P	P	.	.	P	P	P	P	P	P	P	.	.	P	P	P	P	P	.	.	P	P		23	
26	NERPAGARE VAISHALI BAPU	P	P	–	.	P	P	P	P	.	.	P	P	P	P	P	.	.	P	P	P	P	.	.	P	P	.	.	P	P	P		20	
27	NIKAM SUCHITRA JAGANNATH	P	P	P	–	–	P	P	P	P	P	P	P	P	P	P	P	P	P	–	P	P	P	P	–	P	P	P	P	P	P		26	
28	NIKUMBH HARISH RAJESH	P	P	P	P	P	P	P	.	.	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P		26	
29	PARDESHI AKSHAY SANTOSH	P	P	P	P	P	P	P	P	P	.	P	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	P	P		26	
30	PATIL AKASH VIKRAM	P	P	P	P	P	P	.	P	P	P	P	P	P	.	P	P	P	P	P	P	.	P	P	P	.	P	P	.	P	.		24	

31	PATIL AKSHAY VIKAS	P	.	.	.	P	P	P	P	P	P	.	.	P	P	P	.	P	P	P	P	.	.	P	P	P	.	.	P	P	P		20	
32	PATIL ATUL RAJENDRA	P	P	P	P	P	P	.	.	P	P	P	P	.	P	P	P	P	P	P	P	P	P	.	P	P	P	P	P	P	P		26	
33	PATIL DNYANESHWAR ASHOK	P	P	P	P	.	P	P	P	P	P	P	P	P	P	.	P	P	P	P	P	P	P	P	P	P	P	P	.	P	P		27	
34	PATIL MADHURI PANDHARINATH	P	P	P	P	P	P	P	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	P	P	P	P	.	26	
35	PATIL RUPALI DHARMARAJ	P	P	P	P	P	P	P	P	P	P	.	P	P	P	.	P	P	P	.	P	P	P	P	P	P	P	P	P	P	P		27	
36	PATIL SHARDA SAMADHAN	.	P	P	.	.	.	P	P	P	.	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	P		23	
37	PATIL SHITAL SANJAY	P	P	P	P	.	.	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P		23	
38	PATIL SHUBHAM ASHOK	P	P	P	P	P	P	P	P	.	P	P	P	P	P	P	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P		28	
39	PATIL VRUSHALI AVINASH	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	P	P	P	P	.	.	P	P	P	P	P		26	
40	RAJPUT HEMALATA SATISHSING	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	P	P		28	
41	RAJPUT YOGESH ASHOK	P	P	P	P	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	P	.	P	P	P	P	P		27	
42	SALUNKE SHIVANI GANESH	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P		29	
43	SAWALE PAWAN RAJENDRA	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	P	P	P	P	P	P	P	P	P	P	P		26	
44	SONAWANE RINA KADOBA	P	P	P	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P		26	
45	SURALKAR SHUBHAM ASHOK	P	P	P	.	.	P	P	P	.	.	P	P	P	P	.	.	P	P	P	P	P	P	.	.	P	P	P	P	P	P		22	
46	TADAVI RESHAMA SALIM	P	P	P	P	P	P	P	P	P	P	P	P	.	P	P	P	P	P	P	.	P	P	P	P	P	P	P	P	P	P		28	
47	TAYADE GAURAV SANTOSH	P	P	P	.	P	P	P	P	P	P	.	P	P	P	P	P	.	P	P	P	P	P	.	P	P	P	P	P	P	.		25	
48	THORAT AKSHAY PRAKASH																																		
49	THORAT VAIBHAV RAVINDRA	P	P	P	.	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	P	.	.	P	P	P	P		25	
50	CHAUDHARI SALAS RAJENDRA	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	P	P	P	P	P	P	P	P	P		29	
51	CHAVAN NITIN VIJAY	P	P	P	P	P	P	P	.	P	P	P	P	P	.	.	P	P	P	.	.	P	P	P	P	P	P	P	P	.	P		24	
52	DESHMUKH AKSHAY GOPAL	P	P	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	P	P	P		28	
53	GARUD SHUBHAM WAMANRAO	.	P	P	.	.	.	.	.	P	.	.	.	.	.	P	.	.	.	P	P	P	P	P	.	.	.	.	P	P	P		12	
54	GITE MAYURI DEVIDAS	P	P	P	P	P	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		28	
55	GORE PUJA SAMADHAN	.	P	.	P	P	P	P	P	.	P	P	.	P	P	P	P	P	P	P	.	P	P	P	.	.	P	P	P	P	P		23	
56	LAHASE PANKAJ DHANRAJ																																		
57	LOHAR MANGESH KISHOR	P	P	P	P	P	P	P	P	P	.	P	P	P	P	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	.		26	
58	PATEL MARIYA HIFJURRAHEMAN	P	P	P	P	P	P	P	P	P	P	P	P	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.		28	
59	PATIL HARSHAL GOPAL	P	P	.	.	.	.	.	.	P	P	P	P	P	P	P	P	.	.	P	P	P	P	.	P	P	P	P	.	.	.		18	
60	PATIL JAYANT SUBHASH	.	.	.	.	.	.	.	.	P	P	P	P	P	P	P	P	P	.	P	P	P	P	P	P	P	.	P	.	P			19	
61	PATIL MANOJ ASHOK	P	P	P	P	.	.	P	P	P	P	P	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	.	P	P			23	
62	PATIL RAHUL SUNIL	P	P	.	.	.	.	.	.	.	P	P	P	.	.	.	.	.	.	P	P	P	.	.	P	P	P	P	.	.			12	
63	PATIL SHUBHAM DILIP	P	P	P	P	P	.	P	P	P	.	P	P	P	P	P	.	P	P	P	P	P	.	P	P	P	P	P	.	P			24	
64	PAWAR ISHVARKUMAR CHARAN	P	.	.	.	P	P	P	P	P	P	.	.	P	P	P	P	.	P	P	P	P	P	P	P	P	P	P	P	P			24	
65	SHINDE GANESH GOPAL	.	.	.	P	P	P	.	.	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	.	.	P	P	P			19	

SHENDURNI
JAL JAMNER
JALGAON
COLLEGE

66	SURWADE CHANDRAMANI EKNATH																																	
67	SURYAWANSHI POOJA RAJENDRA	P	P	P	P	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	P	P	.	P	P	.	P	P	P	P	.	25	
68	TELI VAIBHAV SURESH	P	.	.	.	P	P	P	.	.	P	P	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	.	P	P	21	
69	VISPUTE NIKEETA RAVINDRA	P	P	P	P	P	P	P	P	P	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	29	

Subject Teacher

Principal

Principal
Appasaheb R. B. Garud Art's
Commerce & Science College
Shendurni-424204, Tal.Janner

Appasaheb Ragunathrao Bhaurao Garud Arts, Commerce and Science College
Pachora Road City: Shendurni, Pin: 424204, Tehsil:Jamner, State:Maharashtra

MARK-LIST (Slow and Advanced Learner) CHEMISTRY

B.Sc. (with Credits)-Regular-CGPA [60+40] Pattern , FY B.Sc. , Sem-I

Academic Year :2015-2016

Division	Roll No	Student Name	Test-1	Test-2
	1	BANKAR BHUSHAN SANJEEV	14	
	2	BANKAR PRADIP SHANTARAM	18	
	3	BHADANE MADHURI RAJENDRA	20	
	4	BHAVSAR JITENDRA PRAMOD	16	
	5	BHOI VIKRAM RAMLAL	22	
	6	BUNDE NAMRATA VASUDEO	10	18
	7	CHAUDHARI KALYANI RAJENDRA	19	
	8	CHAUDHARI NIKITA DINESH	08	24
	9	CHAUDHARI PUJA AMRUT	14	
	10	CHAUDHARI RAHUL NANA	08	23
	11	CHAUDHARI RUPALI GAJANAN	18	
	12	CHAVHAN DIPALI RAJARAM	18	
	13	GHONGADE YOGITA ISHWAR	13	
	14	GUJAR NAINA KISHOR	19	
	15	GUJAR PRIYANKA RAMESH	15	
	16	JADHAV BANDU DEVIDAS	07	21
	17	JADHAV RITESH KESHAV	19	
	18	JANGALE RAJSHRI SANJAY	28	
	19	JOSHI MANASI DILEEP	11	
	20	JOSHI RADHA SHESHRAO	16	
	21	KAKADE SACHIN VASANT	12	
	22	KATE PRATIKSHA PRAKASH	22	
	23	MAHAJAN MAMATA VASANTRAO	14	
	24	MARATHE VIDYA SUKDEV	17	
	25	NAIK SHRIKRUSHNA SUDAM	17	
	26	NERPAGARE VAISHALI BAPU	19	
	27	NIKAM SUCHITRA JAGANNATH	08	24
	28	NIKUMBH HARISH RAJESH	20	
	29	PARDESHI AKSHAY SANTOSH		

Principal
Appasaheb Garud Art's
Commerce and Science College
Shendurni Tal. Jamner

	30	PATIL AKASH VIKRAM	13	
	31	PATIL AKSHAY VIKAS	18	
	32	PATIL ATUL RAJENDRA	18	
	33	PATIL DNYANESHWAR ASHOK	AB	
	34	PATIL MADHURI PANDHARINATH	16	
	35	PATIL RUPALI DHARMARAJ	12	
	36	PATIL SHARDA SAMADHAN	17	
	37	PATIL SHITAL SANJAY	21	
	38	PATIL SHUBHAM ASHOK	AB	
	39	PATIL VRUSHALI AVINASH	AB	
	40	RAJPUT HEMALATA SATISHSING	26	
	41	RAJPUT YOGESH ASHOK	15	
	42	SALUNKE SHIVANI GANESH	23	
	43	SAWALE PAWAN RAJENDRA	30	
	44	SONAWANE RINA KADOBA	28	
	45	SURALKAR SHUBHAM ASHOK	16	
	46	TADAVI RESHAMA SALIM	19	
	47	TAYADE GAURAV SANTOSH	AB	
	48	THORAT AKSHAY PRAKASH	AB	
	49	THORAT VAIBHAV RAVINDRA	12	
	50	CHAUDHARI SALAS RAJENDRA	AB	
	51	CHAVAN NITIN VIJAY	29	
	52	DESHMUKH AKSHAY GOPAL	16	
	53	GARUD SHUBHAM WAMANRAO	AB	
	54	GITE MAYURI DEVIDAS	18	
	55	GORE PUJA SAMADHAN	22	
	56	LAHASE PANKAJ DHANRAJ	29	
	57	LOHAR MANGESH KISHOR	19	
	58	PATEL MARIYA HIFJURRAHEMAN	17	
	59	PATIL HARSHAL GOPAL	12	
	60	PATIL JAYANT SUBHASH	12	
	61	PATIL MANOJ ASHOK	24	
	62	PATIL RAHUL SUNIL	16	
	63	PATIL SHUBHAM DILIP	19	
	64	PAWAR ISHVARKUMAR CHARAN	16	
	65	SHINDE GANESH GOPAL	17	
	66	SURWADE CHANDRAMANI EKNATH	AB	

Appasaheb ... Commerce & ... Shendurni-42 ... Arts College ... Jamner

SECONDARY EDUCATION ... ARTS COMM. & SCIENCE ... SHENDURNI ... TAL JAMNER ... DIST. JALGAON

COLLEGE

SEAT NO. CH20KA

WRONG	RIGHT
⊘ ⊘ ◐ ●	⊙ ⊙ ● ⊙

DETAILS OF STUDENT –

NAME: SURALKAR SHUBHAM ASHOK

CLASS: TYBSC

SUBJECT: CHEMISTRY

DATE OF EXAM: 31 08 15

CENTRE OF EXAM: SMRNDURNI

CODE: 170033

TOTAL: 16

Invigilator's Signature

	A	B	C	D
1	○	●	○	○
2	●	○	○	○
3	○	●	○	○
4	○	○	●	○
5	○	●	○	○
6	○	○	●	○
7	○	●	○	○
8	○	○	●	○
9	●	○	○	○
10	○	○	○	●
11	○	●	○	○
12	○	○	○	●
13	●	○	○	○
14	○	○	○	●
15	●	○	○	○
16	○	○	●	○
17	○	○	●	○
18	●	○	○	○
19	○	○	○	●
20	●	○	○	○
21	●	○	○	○
22	○	●	○	○
23	○	●	○	○
24	●	○	○	○
25	○	●	○	○
26	●	○	○	○
27	●	○	○	○
28	●	○	○	○
29	○	●	○	○
30	○	○	○	●
31	○	○	○	○
32	○	○	○	○
33	○	○	○	○
34	○	○	○	○
35	○	○	○	○
36	○	○	○	○
37	○	○	○	○
38	○	○	○	○
39	○	○	○	○
40	○	○	○	○
41	○	○	○	○
42	○	○	○	○
43	○	○	○	○
44	○	○	○	○
45	○	○	○	○
46	○	○	○	○
47	○	○	○	○
48	○	○	○	○
49	○	○	○	○
50	○	○	○	○
51	○	○	○	○
52	○	○	○	○
53	○	○	○	○
54	○	○	○	○
55	○	○	○	○
56	○	○	○	○
57	○	○	○	○
58	○	○	○	○
59	○	○	○	○
60	○	○	○	○
61	○	○	○	○
62	○	○	○	○
63	○	○	○	○
64	○	○	○	○
65	○	○	○	○
66	○	○	○	○
67	○	○	○	○
68	○	○	○	○
69	○	○	○	○
70	○	○	○	○
71	○	○	○	○
72	○	○	○	○
73	○	○	○	○
74	○	○	○	○
75	○	○	○	○
76	○	○	○	○
77	○	○	○	○
78	○	○	○	○
79	○	○	○	○
80	○	○	○	○
81	○	○	○	○
82	○	○	○	○
83	○	○	○	○
84	○	○	○	○
85	○	○	○	○
86	○	○	○	○
87	○	○	○	○
88	○	○	○	○
89	○	○	○	○
90	○	○	○	○
91	○	○	○	○
92	○	○	○	○
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Appasaheb Birnale College of Arts Commerce

CH201SA

ANSWER SHEET

SEM-I EXAMINATION NOVEMBER

SEAT No. CH201SA

• READ INSTRUCTIONS : Please read the instructions carefully before writing. • Please ONLY ONE CIRCLE with a pen. • Do not make any stray marks on the answer sheet. • Rough work must not be done on the answer sheet. • Do not fold the Answer Sheet

WRONG	RIGHT
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DETAILS OF STUDENT –

NAME THORAT VAIBHAV RAVINDRA

CLASS FYBSC

SUBJECT CHEMISTRY

DATE OF EXAM 31 08 15

CENTRE OF EXAM Shendurni

CODE 170033

TOTAL
12

Invigilator's Signature

	A	B	C	D
01	●	O	O	O
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DEPARTMENT OF ... SCIENCE
SHENDURNI
A.R.S. SABOO COLLEGE

Principal
Appasaheb Birnale Arts's
Commerce and Science College
Shendurni Dist. Kolhapur

A. R. B. Garud Art's, Commerce and Science College, Shendurni.

Tal- Jamner, Dist- Jalgaon. Pin- 424204

Test for Chemistry

30

Select an appropriate option for each of the following

1) The ideal gas equation for 1 mole of gas is----
A] $PV = nRT$ B] $\frac{PV}{nT} = R$ C] $PV = RT$ D] None

2) Which of the following mostly account for the mass of an atom?
A] Neutrons B] Neutrons & Protons C] Electron & Proton D] Electron & Neutron

3) Saturated hydrocarbon are known as
A] Alkanes B] Alkenes C] Alkynes D] Alkaloids

4) The Van der Waal equation of gas is---
A] $PV = nRT$ B] $\left(P+\frac{a}{V^2}\right)(V-nb)= nRT$ C] $\left(P+\frac{a}{V^2}\right)(V-nb^2)= nRT$ D] None

5) Which metal was used by Rutheford in his α- scattering experiment
A] Gold B] Platinum C] Silver D] Lead

6) Identify the smallest alkane
A] Cyclomethane B] Methane C] Cyclopropane D] Propane

7) Van der Waal's equation explains behavior of –
A] Ideal gas B] Real gas C] Nobel gas D] liquefaction of gas

8) An element has an electronic configuration of (2,8,7) it's valency is –
A] 2 B] 7 C] 3 D] 1

9) Dienes are the name given to compounds with –
A] Exactly a double bond B] Exactly a triple bond
C] Exactly two double bond D] More than two double bond

10) Critical temperature of gas is the ___
A] Temp. above which a liquid can not exist B] Temp. below which a liquid can not exist
C] Minimum temp. at which a gas can be liquefied D] Maximum temp. at which a gas can be liquefied

11) During a chemical reaction, atomic number ---
A] Changes B] Remains same
C] Changes and then is restored D] Changes alternately

12) The substituent in the chain is named by replacing the 'ane' in the alkane by ---
A] ene B] ic
C] one D] yl

13) The gas will show ideal beahaviour at –
A] Low temp. and low pressure B] Low temp. and high pressure
C] High temp. and low pressure D] High temp. and high pressure

14) Which isotope of H contains only one proton and no neutron in its nucleus?
A] Protium B] Deuterium
C] Tritium D] None of the above

15) Which of the following is used for artificial ripening of fruit-
A] Ethylene B] Acetylene C] Phenol D] Methanol

16) The gas which shows max. deviation from ideal behavior is ----
A] CO_2 B] O_2 C] He D] N_2

17) The electronic configuration of an atom having atomic number 20 is –
A] (2,8,8,2) B] (2,8,10) C] (2,6,8,4) D] (2,4,8,6)

18) General formula for amine is –
A] $R\text{-}CH_2$ B] R_2 - CH C] R – NH_2 D] R - COOH

19) 'When a gas under high pressure is allow to expand adiabatically through a fine hole into a region of low pressure is accompanied by cooling' , the effect is known as-
A] Joule-Thomson effect B] Linde's effect
C] Temperature effect D] None

20) An atom differs from its ion in
A] Nuclear charge B] Mass number C] Number of electrons D] Number of Neutrons

21) What is the general formula of alkenes-
A] C_nH_{2n+2} B] C_nH_{2n} C] C_nH_{2n-2} D] None

22) The critical temperature for CO_2 gas is ---- ^oC
A] 32 ^oC B] 13.1 ^oC C] 48 ^oC D] 31.1 ^oC

23) Which of the following is not fundamental particle-
A] Proton B] Neutron C] Alpha particle D] Electron

24) Which of the following shows molecular formula of methane-
A] CH_4 B] C_2H_6 C] C_3H_8 D] C_4H_{10}

25) The compressibility factor at the same temperature and pressure is given by—
A] $Z= \frac{PV}{nRT}$ B] $Z= \frac{V_{real}}{V_{ideal}}$ C] Z= nPVR D] None

26) Atom which have same mass number but different atomic number are called ---
A] Isotopes B] Isomer C] Isotones D] Isobars

27) An alkane with Molecular Formula C_7H_{16} is ---
A] Butane B] Pentane C] Hexane D] Heptane

28) If Z > 1 the gas is ---
A] More compressible B] Not compressible C] Less compressible D] None

29) How many electron does the outermost shell of Neon have?
A] 4 B] 8 C] 2 D] 16

30) Alcohols (CnH2n+1OH) are formed by attachment of
A] –OH group to Hydrocarbon chain B] H-X group to Hydrocarbon chain
C] O to Hydrocarbon chain D] H to Hydrocarbon chain

BEST OF LUCK

A. R. B. Garud Art's, Commerce and Science College, Shendurni.
Tal- Jamner, Dist- Jalgaon. Pin- 424204

ANSWER KEY

[Test For Chemistry]

Question No.	Responce	Question No.	Responce
1	C	16	A
2	B	17	A
3	A	18	C
4	B	19	A
5	A	20	C
6	B	21	B
7	B	22	D
8	D	23	C
9	C	24	A
10	D	25	B
11	B	26	D
12	D	27	D
13	C	28	C
14	A	29	B
15	B	30	A

PAPER SETTER
1] MR. S. E. DRAKSHE

2] MR. R. S. GARUD

Principal
A. R. B. Garud's
Commerce and Science College
Shendurni

ANSWER SHEET

(SLOW AND ADVANCED LEARNER)

NO. - CH 2815 A

INSTRUCTIONS : Please read the instruction carefully before writing. • Use BLACK BALL PEN for writing the Roll number, Question paper code and Answer. • Darken ONLY ONE CIRCLE with BLACK BALL PEN for each answer as shown in the examples. If you darken more than one circle your answer will be wrong. • Do not make any stray marks on the answer sheet. • Rough work must not be done on the Answer Sheet. • Do not fold the Answer Sheet.

WRONG	RIGHT
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DETAILS OF STUDENT –

NAME: CHAUDHARI NIKITA R

CLASS: BYJS C

SUBJECT: Chemistry

DATE OF EXAM: 31 08 15

SHIFT OF EXAM: Shchdurni

CODE: 170093

TOTAL: 8

Invigilator's Signature

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Application
Comments
Signature

Appasaheb Ragunathrao Bhaurao Garud Arts, Commerce and Science College, Shendurni.

Pre/Post Test for bridge course

Slow And Advance learner

Subject: Chemistry **Academic Year :2016-2017** **Class: F. Y. B. Sc.** **Month:** September

Roll No	Student Name	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	Total	Sign Of Student
1	AUTI GAYATRI KAILAS	P	P	P	P	P	P	.	.	.	P	P	P	.	P	P	.	P	P	P	P	.	.	P	P	P	P	P	.	.	P		21	
2	BANKAR ARATI SUDHAKAR	P	P	P	P	-	.	.	P	P	P	P	P	P	P	P	P	P	.	P	P	P	P	-	.	P	P	P	P	P	P		24	
3	BARI GAURAV DIPLAL	P	P	P	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		27	
4	BARI MANGESH DHANRAJ	—	P	P	P	P	P	P	P	P	P	P	P	P	.	.	P	P	P	P	P	-	.	P	P	P	P	P	P	P	P		25	
5	BARI RAJESH SUDHAKAR	P	P	P	P	P	P	-	.	P	P	P	P	P	P	.	P	P	P	P	P	P	P	P	.	.	P	P	P	P	P		25	
6	BASNIWAL JYOTI JAGMOHANLAL .	P	—																														01	
7	BAVISKAR PANKAJ RAJU	P	P	P	P	P	P	P	-	.	.	P	P	P	P	P	P	.	P	P	P	P	.	.	.	P	P	P	.	.	P		21	
8	BORASE VAIBHAV BHAGAWAT	P	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	-	.	P	P	P	P	P	-	-	-	P	P	-	P		22	
9	CHAUDHARI PRAGATI RAMESH	-	P	P	P	P	P	P	P	P	P	P	.	-	.	P	P	P	P	-	P	-	P	-	P	P	P	P	.	P	P		22	
10	CHAUDHARI SAKSHI SANJAY	P	P	P	—	P	-	P	P	P	P	P	P	P	—	P	-	—	P	P	P	P	P	P	P	P	—	P	P	P	P		24	
11	CHAUTHE ASHUTOSH SANTOSH	P	P	P	P	P	P	P	-	P	—	P	P	P	P	P	P	P	P	—	—	P	P	P	P	P	P	P	P	P	P		26	
12	DESHMUKH SHITAL VIJAY	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	—	—	—	P	P	P	P	—	P	P	P	P	P	P	P		26	
13	GANDHILE SHITAL DINKAR	P	P	—	P	P	P	P	P	P	P	—	P	P	P	P	—	P	P	P	P	P	P	—	P	P	P	-	-	P	P		24	
14	GARUD YOGITA NIVRUTTI	P	P	P	P	P	—	P	P	P	P	P	P	.	P	P	P	P	-	P	P	P	P	—	P	P	P	P	P	P	P		26	
15	GHATE HARSHALA ATMARAM	P	P	.	.	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	.	P	P	P	P	P	.	.	P	P	P		23	
16	GHULE HARSHAL PRATAP	P	P	P	P	—	—	P	P	P	P	P	P	—	—	P	P	P	P	P	P	—	P	P	P	P	—	P	P	P	P		24	
17	GITE FANESHWAR SUNIL	P	P	P	P	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	.		24	
18	GOPAL RAVINDRA HIRALAL KISAN GOPAL	P	P	P	P	P	P	.	.	P	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		26	
19	GUJAR ANKESH SHARAD	P	P	P	P	P	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P	.	.	P	P	P	P		25	
20	GUJAR JAGRUTI KALYAN .	—																																
21	HATKAR ATUL BAPU	P	P	P	P	P	P	P	P	-	P	P	P	P	P	P	P	P	-	P	P	P	P	P	P	-	P	-	P	P	P		26	
22	IKHANKAR OJAS ISHWAR	P	P	P	—	P	P	P	P	P	P	P	—	P	P	P	P	P	P	—	P	P	P	P	—	P	P	P	-	P	P		25	
23	JADHAV DIPAK DEVRAM .	—																																
24	JOSHI VIDYA HARI	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		25	
25	KALE KOMAL VIJAY	P	P	P	.	.	P	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.		25	
26	KAPSE NIKITA VIJAY	.	.	P	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		26	
27	KAPURE YASHAWANT BHAGWAT	P	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P	.	.	P	P	P	.	.	.	P	.	.	P		21	
28	KHATIK DANISH SHAKIL	.	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	P	P	-	P	P	P	P	.	P	P		25	

29	KHATIK SALMAN FAROOQUE	P	P	P	P	–	–	–																											94
30	KOLHE TUSHAR DNYANESHWAR	P	P	P	–	–	P	P	P	P	P	P	–	P	P	P	P	P	P	P	P	P	–	P	P	P	P	–	–	P	P		25		
31	KOLI SADHANA NANA	P	P	P	P	P	P	–	–	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		28		
32	MALI KIRAN KAILAS	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		30		
33	MALI SACHIN ASHOK	P	P	P	P	P	P	P	P	P	P	P	P	P	–	–	–	P	P	P	P	P	P	P	P	P	P	P	P	P	P		27		
34	MALKAR RUPESH SUDHAKAR	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		29		
35	NIKAM YOGESH AVINASH	–	P	P	P	P	P	P	–	–	P	P	P	P	P	P	P	P	P	P	P	P	–	–	P	P	P	P	–	P			24		
36	NIKUMBH SHUBHAM PAVAN																																		
37	PARDESHI SAGAR DNYANESHWAR	–	P	P	P	P	P	P	P	P	P	P	P	P	–	–	P	P	P	P	P	P	P	P	–	P	P	P	–	–	P		24		
38	PARDESHI YOGESH ASHOK	P	P	P	P	–	–	–	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		27		
39	PATHAN MOBINKHA AKBARKHA	P	P	P	P	P	P	P	P	P	P	–	–	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		28		
40	PATIL GANESH EKNATH	P	–	–	P	P	P	P	P	P	P	–	P	P	P	P	P	P	P	P	–	–	P	P	–	–	P	P	P	P	P		23		
41	PATIL MAYA RAVINDRA	P	P	P	P	P	P	–	–	P	P	P	P	P	P	P	–	P	P	P	P	P	P	P	–	–	P	P	P	P	P		25		
42	PATIL MAYUR JANKIRAM	P	P	P	–	–	P	P	P	P	P	P	–	P	P	P	P	P	–	P	P	P	–	P	P	P	–	–	P	P	P		23		
43	PATIL PRADIP SANTOSH	P	P	P	P	P	P	–	–	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		28		
44	PATIL PUNAM DINESH	P	P	P	P	P	P	P	P	P	P	–	–	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		28		
45	PATIL RENUKA BHARAT	P	P	P	P	P	–	P	P	P	–	P	P	–	P	P	P	P	P	–	P	P	P	P	P	P	P	P	P	P	P		26		
46	PATIL SACHIN ATMARAM	P	P	P	–	P	P	–	P	P	P	P	P	P	P	P	P	–	P	P	P	P	P	P	–	–	P	P	–	P	P		24		
47	PATIL SHASHIKAT BHASKAR	P	P	P	P	P	P	P	P	–	–	P	P	P	P	P	P	–	P	P	P	P	–	P	P	–	P	P	P	P	P		25		
48	PATIL SHUBHAM GOVINDA	P	P	–	P	P	P	P	P	P	P	P	P	–	–	–	P	P	P	P	P	P	P	P	P	P	P	–	P	P	–		24		
49	PATIL SHUBHANGI SUNIL	P	P	P	P	P	P	P	–	P	P	P	P	P	P	P	P	P	P	P	P	P	P	–	P	P	P	P	P	P	P		28		
50	PATIL VAISHALI SANJAY	P	P	P	P	P	P	P	P	P	P	P	P	P	–	P	P	–	P	P	P	P	P	P	P	P	P	P	P	P	P		28		
51	PATIL VAISHNAVI RAVINDRA	P	P	P	P	P	P	P	P	P	P	–	P	P	P	–	–	P	P	P	–	–	–	–	P	P	P	P	P	P	P		23		
52	PAWAR RITESH VISHNU																																		
53	RATHOD MUKESH SUDAM	P	P	P	P	P	P	P	–	P	P	P	P	–	–	P	P	P	P	P	P	–	–	P	P	P	P	P	P	P	P		25		
54	RATHOD NITESH NANU	P	P	P	P	P	P	P	P	P	P	–	P	P	P	P	–	–	P	–	–	–	P	P	P	–	P	P	–	P	P		22		
55	RAUT SHIVESH SHANKAR	P	P	P	P	P	P	P	P	–	P	P	–	P	P	P	P	P	P	P	P	P	P	–	P	P	P	P	P	P	P		27		
56	SANDANSHIV PRASHIK RAJENDRA	P	P	–	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	–	P	P	P	P	P	P		28		
57	SAVALE VIKAS BABURAO ANADA SAVALE	P	P	P	P	–	P	–	P	P	P	–	–	–	P	P	–	P	P	P	P	P	P	P	P	P	–	P	P	P	P		23		
58	SHAIKH DANISH SHAMSUDDIN	P	P	P	–	–	P	P	P	P	P	P	P	P	P	P	–	P	P	P	–	P	P	P	–	P	–	P	–	P	P		23		
59	SISODE GAURAVKUMAR KOMALSING	P	P	P	P	–	P	P	P	P	P	–	P	P	P	P	–	P	P	P	P	P	P	P	–	–	P	P	–	P	P	P	24		
60	SONAWANE SUDIKSHA ASHOK	P	P	P	–	P	P	P	P	P	–	–	–	P	P	P	P	P	P	P	P	P	P	–	–	P	P	P	P	P	P	–	23		
61	SURYAWANSHI AKASH KIRAN	P	P	P	P	P	P	–	P	P	P	P	P	P	P	P	P	P	–	P	P	P	P	P	P	P	–	P	P	–	P		26		

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62	SURYAWANSHI PARMESHWAR SURESH	P	P	P	-	-	-	-	P	P	P	P	P	-	-	-	-	P	-	P	-	P	-	P	P	P	P	P	P	P	P		19
63	TELI SACHIN JIVAN	-	-	-	P	P	P	P	P	-	-	-	P	P	P	-	-	-	P	P	P	P	P	-	P	P	P	P	P	P	P		20
64	THORAT DHIRAJ SUDHAKAR	P	P	P	P	-	-	P	P	P	P	P	P	-	-	P	P	P	P	P	P	-	P	P	P	P	P	P	P	P	P		25
65	WAGH VIVEK VASANT	P	P	P	P	P	P	P	P	P	P	P	P	P	P	-	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P		28
66	BANKAR HITENDRA VIJAY	P	P	-	-	-	-	-	P	P	-	-	-	-	-	P	P	P	P	P	-	-	P	P	P	P	P	P	P	P	P		18
67	CHAUDHARI ROSHAN GANESH	P	P	P	P	P	P	P	P	P	P	P	P	P	P	-	-	-	P	P	P	P	P	-	-	-	P	P	P	P	P		24
68	CHAUDHARI SONALI DNYANESHWAR	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	-	P	-	-	-	-	-	P	P	-	-	-	-	-	P		19
69	CHAVAN BHAGYASHRI RANJIT	P	P	P	-	-	-	-	-	P	P	P	P	-	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		22
70	DESHMUKH PRASHANT BALU	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	-	-	-	P	P	P	P	P	P	P	P	P	P	P		27
71	GARUD AKASH MANOJ	P	P	P	P	P	P	P	P	P	P	-	-	P	P	P	P	P	P	P	-	-	P	-	P	P	P	P	P	P	P		25
72	GARUD PRANAV SANJAY	P	P	P	P	P	-	P	P	P	P	P	P	P	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		28
73	GITE MAYUR SANJAY	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	-	P	P	P	P	P	P	P	P	P	P	P	P		29
74	GITE SOPAN ISHWAR	P	P	-	P	P	-	P	-	P	P	-	P	P	-	P	P	-	P	-	P	-	P	-	P	-		-	P	P	P		19
75	JADHAV RAJU RAMDAS	P	P	P	P	P	P	P	-	P	P	-	P	-	P	P	-	P	P	P	P	-	P	P	-	P	P	-	P	-	P		22
76	KHAN ARBAZ AQEEL	-	-	P	P	P	-	-	P	P	P	P	-	P	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		24
77	LAHASE RAJESH VILAS	P	P	P	P	P	P	P	-	-	P	P	P	P	P	P	-	-	P	P	P	P	P	P	P	P	P	P	P	P	P		26
78	NILE BHAVESH ASHOK	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	-	P	P	P	-	P	P	P		28
79	PATIL SATISH AMRUT	P	P	P	P	P	P	P	P	P	P	P	P	-	P	P	P	-	P	P	P	-	P	P	P	P	P	P	P	P	P		27
80	PATIL BHUSHAN SUBHASH	P	P	P	P	P	P	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		29
81	PATIL GOVINDA VIJAY	P	P	P	P	P	P	P	P	P	-	P	-	P	P	-	P	-	P	P	-	P	-	P	-	P	-	P	-	P	-		20
82	PATIL ISHWAR ABHIMAN	P	P	-	P	P	P	-	P	-	P	P	-	P	P	-	P	P	-	P	P	P	P	P	P	P	P	P	P	P	P		24
83	PATIL MADHURI VIKAS	P	P	P	P	-	P	P	P	P	P	P	P	-	P	P	P	P	P	P	-	P	P	P	-	P	-	P	-	P	-		23
84	PATIL MANGESH JANARDHAN	P	-	P	P	P	P	-	P	P	-	P	P	P	P	P	P	-	-	P	P	P	P	P	P	P	P	P	P	P	P		25
85	PATIL PRAGATI KISHOR	P	P	P	P	P	P	P	P	P	P	P	P	-	P	P	P	P	P	P	-	P	P	P	-	P	P	P	P	P	P		27
86	PATIL RAHUL DEEPAK	P	P	P	P	P	P	-	P	P	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		28
87	PATIL RUPESH VIJAY	P	P	P	P	P	P	P	P	P	P	P	P	P	P	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		29
88	PATIL TUSHAR ANIL	P	P	P	P	P	P	P	-	P	P	P	-	P	P	P	-	P	P	P	-	P	P	-	P	P	-	P	P	P	P		24
89	PATIL VASUDEV BHAGWAN	-	P	P	P	P	-	P	P	P	-	P	P	P	-	P	P	P	-	P	P	-	P	P	-	P	P	P	-	P	P		22
90	PAWAR UMESH CHHAGAN	P	P	P	-	P	P	P	-	P	P	P	P	-	P	P	P	P	P	P	-	P	P	P	P	P	P	-	P	P	P		25
91	SABALE DEEPAK KAILAS	P	P	P	P	P	-	P	P	P	P	-	P	P	P	P	P	-	P	P	P	P	P	P	P	P	P	P	P	P	P		27
92	SHELAKE VISHAL DILEEP	P	P	-	P	P	P	P	P	P	P	P	P	P	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		28
93	SONAWANE SUVARNA ASHOK	P	P	P	P	P	P	P	P	P	P	P	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		29

Subject Teacher

Principal

Appasaheb Garud Art's Commerce & Science College Shendurni-424203, Tal. Jamner

Appasaheb Ragunathrao Bhaurao Garud Arts, Commerce and Science College

Pachora Road City: Shendurni, Pin: 424204, Tehsil:Jamner, State:Maharashtra

MARK-LIST (Slow and Advanced Learner) CHEMISTRY

B.Sc. (with Credits)-Regular-CGPA [60+40] Pattern , FY B.Sc. , Sem-I

Academic Year :2016-2017

Division	Roll No	Student Name	Test-1	Test-2
	1	AUTI GAYATRI KAILAS	08	25
	2	BANKAR ARATI SUDHAKAR	29	~~27~~
	3	BARI GAURAV DIPLAL	16	
	4	BARI MANGESH DHANRAJ	17	
	5	BARI RAJESH SUDHAKAR	AB	
	6	BASNIWAL JYOTI JAGMOHANLAL	AB	
	7	BAVISKAR PANKAJ RAJU	07	19
	8	BORASE VAIBHAV BHAGAWAT	27	28
	9	CHAUDHARI PRAGATI RAMESH	13	
	10	CHAUDHARI SAKSHI SANJAY	21	22
	11	CHAUTHE ASHUTOSH SANTOSH	14	
	12	DESHMUKH SHITAL VIJAY	15	
	13	GANDHILE SHITAL DINKAR	AB	
	14	GARUD YOGITA NIVRUTTI	13	
	15	GHATE HARSHALA ATMARAM	21	
	16	GHULE HARSHAL PRATAP	14	
	17	GITE FANESHWAR SUNIL	09	23
	18	GOPAL RAVINDRA HIRALAL KISAN GOPAL	18	18
	19	GUJAR ANKESH SHARAD	05	20
	20	GUJAR JAGRUTI KALYAN	12	
	21	HATKAR ATUL BAPU	18	~~18~~
	22	IKHANKAR OJAS ISHWAR	18	
	23	JADHAV DIPAK DEVRAM	19	
	24	JOSHI VIDYA HARI	05	25
	25	KALE KOMAL VIJAY	08	20
	26	KAPSE NIKITA VIJAY	07	27
	27	KAPURE YASHAWANT BHAGWAT		

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	28	KHATIK DANISH SHAKIL	19	
	29	KHATIK SALMAN FAROOQUE	AB	
	30	KOLHE TUSHAR DNYANESHWAR	18	
	31	KOLI SADHANA NANA	13	
	32	MALI KIRAN KAILAS	11	
	33	MALI SACHIN ASHOK	16	
	34	MALKAR RUPESH SUDHAKAR	15	
	35	NIKAM YOGESH AVINASH	17	
	36	NIKUMBH SHUBHAM PAVAN	17	
	37	PARDESHI SAGAR DNYANESHWAR	14	
	38	PARDESHI YOGESH ASHOK	18	
	39	PATHAN MOBINKHA AKBARKHA	18	
	40	PATIL GANESH EKNATH	16	
	41	PATIL MAYA RAVINDRA	16	
	42	PATIL MAYUR JANKIRAM	16	
	43	PATIL PRADIP SANTOSH	16	
	44	PATIL PUNAM DINESH	19	
	45	PATIL RENUKA BHARAT	22	
	46	PATIL SACHIN ATMARAM	16	
	47	PATIL SHASHIKAT BHASKAR	15	
	48	PATIL SHUBHAM GOVINDA	16	
	49	PATIL SHUBHANGI SUNIL	15	
	50	PATIL VAISHALI SANJAY	16	
	51	PATIL VAISHNAVI RAVINDRA		
	52	PAWAR RITESH VISHNU	15	
	53	RATHOD MUKESH SUDAM	17	
	54	RATHOD NITESH NANU	16	
	55	RAUT SHIVESH SHANKAR	7	
	56	SANDANSHIV PRASHIK RAJENDRA	14	
	57	SAVALE VIKAS BABURAO ANADA SAVALE	14	
	58	SHAIKH DANISH SHAMSUDDIN	15	
	59	SISODE GAURAVKUMAR KOMALSING	12	
	60	SONAWANE SUDIKSHA ASHOK	14	
	61	SURYAWANSHI AKASH KIRAN	13	

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Shendurni Tal. Jamner

	62	SURYAWANSHI PARMESHWAR SURESH	17	
	63	TELI SACHIN JIVAN	17	
	64	THORAT DHIRAJ SUDHAKAR	17	
	65	WAGH VIVEK VASANT	17	
	66	BANKAR HITENDRA VIJAY	27	
	67	CHAUDHARI ROSHAN GANESH	16	
	68	CHAUDHARI SONALI DNYANESHWAR	17	
	69	CHAVAN BHAGYASHRI RANJIT	17	
	70	DESHMUKH PRASHANT BALU	17	
	71	GARUD AKASH MANOJ	17	
	72	GARUD PRANAV SANJAY	13	
	73	GITE MAYUR SANJAY	15	
	74	GITE SOPAN ISHWAR	17	
	75	JADHAV RAJU RAMDAS	16	
	76	KHAN ARBAZ AQEEL	16	
	77	LAHASE RAJESH VILAS	15	
	78	NILE BHAVESH ASHOK	13	
	79	PATIL SATISH AMRUT	15	
	80	PATIL BHUSHAN SUBHASH	13	
	81	PATIL GOVINDA VIJAY	16	
	82	PATIL ISHWAR ABHIMAN	17	
	83	PATIL MADHURI VIKAS	14	
	84	PATIL MANGESH JANARDHAN	17	
	85	PATIL PRAGATI KISHOR	18	
	86	PATIL RAHUL DEEPAK	17	
	87	PATIL RUPESH VIJAY	17	
	88	PATIL TUSHAR ANIL	17	
	89	PATIL VASUDEV BHAGWAN	15	
	90	PAWAR UMESH CHHAGAN	15	
	91	SABALE DEEPAK KAILAS	15	
	92	SHELAKE VISHAL DILEEP	17	
	93	SONAWANE SUVARNA ASHOK	16	

Principal
Arts Garud Art's
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ANSWER SHEET

TEST NO. CH-2016A

IMPORTANT INSTRUCTIONS: Please read the instruction carefully before writing. • USE ONLY BALL PEN to write the test number, Question paper code and Answer. • Darken ONLY ONE CIRCLE with BLACK BALL PEN to mark your answer as shown in the Example. If you darken more than one circle your answer will be treated as wrong. • Do not make any stray marks on the answer sheet. • Rough work must not be done on the answer sheet. • Do not fold the Answer Sheet.

WRONG | RIGHT

DETAILS OF STUDENT –

NAME: RATHOD NINESH RANU

CLASS: FYBSC

SUBJECT: CHEMISTRY

DATE OF EXAM: 22 08 16

CENTRE OF EXAM: SHENDURNI

CODE: 170083

TOTAL: 16

Invigilator's Signature

	A	B	C	D
1	●	O	O	O
2	●	O	O	O
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9	O	●	O	O
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ANSWER SHEET

(ANSWER ONLY)

INSTRUCTIONS

- Read the instructions carefully before writing.
- Enter ONE CIRCLE with your answer.
- Rough work must not be done.
- Do not make any stray marks on the answer sheet.
- Do not fold the Answer Sheet.

WRONG | RIGHT

DEPARTMENT OF COMPUTER SCIENCE — SHEVGAON — A.R.B.C.A. COLLEGE

DETAILS OF STUDENT –

NAME: SADASHIV PRASHK DAJENDRA

CLASS: FYBSC

SUBJECT: CHMESTRY

DATE OF EXAM: 22 08 16

CENTRE OF EXAM: SHENDURNI

CODE: 170033

TOTAL: 14

Invigilator's Signature

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2	●	O	O	O
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12	●	O	O	O
13	O	O	●	O
14	O	●	O	O
15	O	O	O	●
16	●	O	O	O
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29	O	O	●	O
30	●	O	O	O
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32	O	O	O	O
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100	O	O	O	O

Principal
Arts, Science and Art's
Commerce & Science College
Shendurni Road, Tal. Jamner

A. R. B. Garud Art's, Commerce and Science College, Shendurni,

Tal- Jamner, Dist- Jalgaon. Pin- 424204

Test for Chemistry

30

Select an appropriate option for each of the following

1) The ideal gas equation for 1 mole of gas is----
A] $PV = nRT$ B] $\frac{PV}{nT} = R$ C] $PV = RT$ D] None

2) Which of the following mostly account for the mass of an atom?
A] Neutrons B] Neutrons & Protons C] Electron & Proton D] Electron & Neutron

3) Saturated hydrocarbon are known as
A] Alkanes B] Alkenes C] Alkynes D] Alkaloids

4) The Van der Waal equation of gas is---
A] $PV = nRT$ B] $\left(P+\frac{a}{V^2}\right)(V-nb) = nRT$ C] $\left(P+\frac{a}{V^2}\right)(V-nb^2) = nRT$ D] None

5) Which metal was used by Rutheford in his α- scattering experiment
A] Gold B] Platinum C] Silver D] Lead

6) Identify the smallest alkane
A] Cyclomethane B] Methane C] Cyclopropane D] Propane

7) Van der Waal's equation explains behavior of –
A] Ideal gas B] Real gas C] Nobel gas D] liquefaction of gas

8) An element has an electronic configuration of (2,8,7) it's valency is –
A] 2 B] 7 C] 3 D] 1

9) Dienes are the name given to compounds with –
A] Exactly a double bond B] Exactly a triple bond
C] Exactly two double bond D] More than two double bond

10) Critical temperature of gas is the ___
A] Temp. above which a liquid can not exist B] Temp. below which a liquid can not exist
C] Minimum temp. at which a gas can be liquefied D] Maximum temp. at which a gas can be liquefied

11) During a chemical reaction, atomic number ---
A] Changes B] Remains same
C] Changes and then is restored D] Changes alternately

12) The substituent in the chain is named by replacing the 'ane' in the alkane by ---
A] ene B] ic
C] one D] yl

13) The gas will show ideal beahaviour at –
A] Low temp. and low pressure B] Low temp. and high pressure
C] High temp. and low pressure D] High temp. and high pressure

14) Which isotope of H contains only one proton and no neutron in its nucleus?
A] Protium B] Deuterium
C] Tritium D] None of the above

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Appasaheb N. D. Garud Art's
Commerce & Science College
Shendurni-424204, Tal. Jamner

A. R. B. Garud Art's, Commerce and Science College, Shendurni.
Tal- Jamner, Dist- Jalgaon. Pin- 424204

ANSWER KEY

[Test For Chemistry]

Question No.	Responce	Question No.	Responce
1	C	16	A
2	B	17	A
3	A	18	C
4	B	19	A
5	A	20	C
6	B	21	B
7	B	22	D
8	D	23	C
9	C	24	A
10	D	25	B
11	B	26	D
12	D	27	D
13	C	28	C
14	A	29	B
15	B	30	A

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Commerce & Science College
Shendurni, Tal-Jamner

PAPER SETTER
1| MR. S. E. DRAKSHE

2| MR. R. S. GARUD

ANSWER SHEET
[SLOW AND ADVANCED LEARNER]

ST NO. - CH2016A

RTANT INSTRUCTIONS : Please read the instruction carefully before writing. • Darken ONLY ONE CIRCLE with the ONLY BLACK BALL PEN for writing the Roll number, Question paper code and Answer the BLACK BALL PEN for each answer as shown in the example. If you darken more than one circle your answer be treated as wrong • Do not make any stray marks on the answer sheet • Rough work must not be done the answer sheet • Do not fold the Answer Sheet

WRONG ⊗ ⊘ ◐ ◑

RIGHT ⊙ ⓑ ● ⓓ

TAILS OF STUDENT –

AME BAVISKAR PANKAJ RAJU

ASS FYBSC

BJECT CHEMISTRY

ATE OF EXAM 22082016

ENTRE OF EXAM SHENDURNI

CODE 170033

TOTAL 07/30

Invigilator's Signature

	A	B	C	D
1	O	O	●	O
2	●	O	O	O
3	O	O	●	O
4	O	O	●	O
5	O	●	O	O
6	O	O	O	●
7	O	O	O	●
8	O	O	●	O
9	O	O	●	O
10	●	O	O	O
11	O	O	●	O
12	O	O	O	●
13	O	●	O	O
14	●	O	O	O
15	O	O	●	O
16	O	●	O	O
17	O	O	●	O
18	●	O	O	O
19	O	●	O	O
20	O	●	O	O

	A	B	C	D
21	O	●	O	O
22	O	●	O	O
23	O	O	●	O
24	●	O	O	O
25	O	O	O	●
26	O	O	●	O
27	O	●	O	O
28	●	O	O	O
29	O	O	●	O
30	O	O	O	●
31	O	O	O	O
32	O	O	O	O
33	O	O	O	O
34	O	O	O	O
35	O	O	O	O
36	O	O	O	O
37	O	O	O	O
38	O	O	O	O
39	O	O	O	O
40	O	O	O	O

	A	B	C	D
41	O	O	O	O
42	O	O	O	O
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45	O	O	O	O
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90	O	O	O	O
91	O	O	O	O
92	O	O	O	O
93	O	O	O	O
94	O	O	O	O
95	O	O	O	O
96	O	O	O	O
97	O	O	O	O
98	O	O	O	O
99	O	O	O	O
100	O	O	O	O

Principal
Garud Art's
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Appasaheb Ragunathrao Bhaurao Garud Arts, Commerce and Science College
Pachora Road City: Shendurni, Pin: 424204, Tehsil:Jamner, State:Maharashtra

MARK-LIST (Slow and Advanced Learner) CHEMISTRY

B.Sc. (with Credits)-Regular-CGPA [60+40] Pattern , FY B.Sc. , Sem-I

Academic Year :2017-2018

Division	Roll No	Student Name	Test-1	Test-2
	1	BANKAR SHASHIKANT PRAKASH	16	
	2	BARI BHAGYASHRI ATMARAM	06	24
	3	BARI VAIBHAV SUNIL	17	
	4	BAVISKAR KAVITA YUVRAJ	16	
	5	BELPATRE RUSHIKESH NILKANTH	15	
	6	BHADANGE AJAY DAGADU	16	
	7	BIDHE MEGHA RAJENDRA	15	
	8	BORSE SHASHANK AVINASH	15	
	9	CHAUDHARI AISHWARYA GAJANAN	13	
	10	CHAUDHARI AKSHAY DATTU	18	
	11	CHAUDHARI KIRAN VILAS	14	
	12	CHAUDHARI VAIBHAV RAVINDRA	15	
	13	CHAVHAN VANDANA RAMESH	28	
	14	CHIRMADE MOHIT CHANDRAKANT	16	
	15	GAIKWAD KOMAL PRAKASH	27	
	16	GARUD MAMTA KRUSHNA	13	
	17	GARUD NIKHIL SHRIKANT	12	
	18	GAYAKWAD HARIDAS BHASKAR	AB	
	19	GHONGADE KAVITA ISHWAR	11	
	20	GITE VISHAL SANJAY	29	
	21	GUJAR MANISHA RAMESH	16	
	22	GUJAR YUGANDHARA KIRAN	14	
	23	JADHAO POONAM VISHNU	AB	
	24	JADHAV SOMNATH HIMMATRAO	12	
	25	KALE ROHINI KAILAS	16	
	26	KAPRE KOMAL SUBHASH	13	

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	27	KASTURE MEGHA MANOJ	15	
	28	KHAIRNAR KALPESH SANJAY	09	25
	29	KSHIRSAGAR VISHNU ASHOK	AB	
	30	KUMAVAT NITEEN HARI	17	
	31	LINGAYAT KISHOR RAJU	17	
	32	MAHAJAN NILESH SAMADHAN	16	
	33	MALI SAGAR RAMESH	19	
	34	MALI SWATI VASANTA	12	
	35	NAIK AASHISH PRABHU	16	
	36	NIKAM RAVINDRA BHIMRAO	17	
	37	NILPAGARE SACHIN DNYANESHWAR	AB	
	38	PARDESHI ISHWAR JAGDISH	18	
	39	PARDESHI JAYAVANT DILIP	11	
	40	PATEL MANASI TEJAS	13	
	41	PATIL ARATI PANKAJ	15	
	42	PATIL BHAGYASHRI RAJENDRA	AB	
	43	PATIL DIPAK RAJU	15	
	44	PATIL DIVYA CHANDRAKANT	15	
	45	PATIL KUNAL BHAGWAN	14	
	46	PATIL MANOJ BHIKA	16	
	47	PATIL MAYUR VISHWANATH	16	
	48	PATIL PALLAVI SANJAY	10	
	49	PATIL SANJANA BANDU	13	
	50	PATIL SWATI EKNATH	AB	
	51	PATIL VAIBHAV ANANDA	18	
	52	PAWAR MADHURI SATISH	AB	
	53	PAWAR RAHUL RAMESH	11	
	54	PHUSE SHUBHANGI SANJAY	15	
	55	PINJARI SHAKIL KADIR	18	
	56	RATHOD HIMMATSING JAGANNATH	20	
	57	SABANE KALYANI SHIVRAM	20	
	58	SAIYYAD SIMIN AKHTAR NASIR ALI	AB	
	59	SAPKAL SHUBHAM MURLIDHAR	AB	
	60	SHAIKH ARBAAJ YUNUS	AB	
	61	SONET RAJ RUMSING	18	
	62	TADAVI SADIK AYYUB	AB	
	63	USHIR VAIBHAV BHAGWAN	21	

	64	VIKHANKAR VIJAY SUDHAKAR	AB	
	65	WAGH SHUBHAM VILAS	AB	
	66	AAGE HARSHAL YUVARAJ	AB	
	67	BANKAR AKASH VIKAS	AB	
	68	BANKAR MAMATA VITTHAL	AB	
	69	DEKE BHUSHAN VASUDEV	11	
	70	DHANGAR RUPALI KIRAN	AB	
	71	GHOPE KANCHAN NAMDEO	10	27
	72	JADHAV MANOJ MOTILAL	AB	
	73	KHAKARE SNEHAL VILASH	AB	
	74	MALI AKASH SUDHAKAR	17	
	75	MALI PRAMOD KAILASH	AB	
	76	PATIL NILESH HIRALAL	AB	
	77	SHRAWANE ASHWINI SHAMRAO	AB	
	78	SHRUNGARE PRAGATI BHAVADU	AB	
	79	BARI TEJAS NANDAKISHOR		
	80	BAVSKAR SACHIN GUNVANT	27	
	81	BHOI MUKESH SANTOSH	06	27
	82	CHAVAN ABHISHEK PRAKASH	AB	
	83	MALI HEMANT DNYANDEV	AB	
	84	PARDESHI GAJANAN SUVARNASING	AB	
	85	PATHAN SHOHEB MUKTAR	AB	
	86	PAWAR MANOJ RAMDAS	20	
	87	SHAIKH MOHAMMAD TOHID MUSHTAK	AB	
	88	TELI DEVANAND VITTHAL	AB	
	89	THORAT DEEPALI VIJAY	20	

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Appasaheb Ragunathrao Bhaurao Garud Arts, Commerce and Science College, Shendurni.

Pre/Post Test for bridge course
Slow And Advance learner
Academic Year :2017-2018

Subject: Chemistry **Class: F. Y. B. Sc.** **Month:** August

Roll No	Student Name	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	Total	Sign Of Student
1	BANKAR SHASHIKANT PRAKASH	P	P	P	P	P	.	P	P	P	P	P	P	P	P	P	P	-	.	.	P	P	P	.	P	P	P	P	P	P	P	P	26	
2	BARI BHAGYASHRI ATMARAM	P	P	P	P	.	.	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	.	.	P	P	P	P	.	24	Bari
3	BARI VAIBHAV SUNIL	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	P	P	P	P	.	P	P	.	P	24	
4	BAVISKAR KAVITA YUVRAJ	P	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	.	P	P	P	P	P	P	P	P	27	
5	BELPATRE RUSHIKESH NILKANTH	P	P	P	.	P	P	.	.	P	P	.	P	P	P	.	.	P	P	P	P	P	P	.	P	P	P	P	P	P	.	.	22	
6	BHADANGE AJAY DAGADU	P	P	.	.	.	P	P	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	25	
7	BIDHE MEGHA RAJENDRA	P	P	P	P	P	P	P	P	.	.	.	P	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	27	
8	BORSE SHASHANK AVINASH	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	P	.	P	P	.	P	P	.	P	P	P	P	P	.	P	P	24	
9	CHAUDHARI AISHWARYA GAJANAN	P	P	P	P	P	.	.	.	.	P	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	24	
10	CHAUDHARI AKSHAY DATTU	P	P	P	P	P	P	P	P	P	.	.	.	P	P	P	.	.	P	P	P	P	P	P	P	P	P	.	.	P	P	P	24	
11	CHAUDHARI KIRAN VILAS	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	.	P	P	P	P	P	27	
12	CHAUDHARI VAIBHAV RAVINDRA	P	P	P	P	P	P	P	P	P	P	.	.	P	.	.	P	P	P	P	P	P	P	P	P	P	P	P	.	P	P	P	26	
13	CHAVHAN VANDANA RAMESH	P	P	P	.	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	26	
14	CHIRMADE MOHIT CHANDRAKANT	P	P	P	P	P	P	P	P	P	P	P	.	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	26	
15	GAIKWAD KOMAL PRAKASH	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	.	P	P	P	P	P	27	
16	GARUD MAMTA KRUSHNA	P	P	P	P	P	P	P	P	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	27	
17	GARUD NIKHIL SHRIKANT	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	28	
18	GAYAKWAD HARIDAS BHASKAR	P	P	P	.	.	.	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	26	
19	GHONGADE KAVITA ISHWAR	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	28	
20	GITE VISHAL SANJAY	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	28	
21	GUJAR MANISHA RAMESH	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	28	
22	GUJAR YUGANDHARA KIRAN	P	P	P	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	28	
23	JADHAO POONAM VISHNU	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	28	
24	JADHAV SOMNATH HIMMATRAO	P	P	P	P	P	P	P	P	.	P	P	P	.	.	P	P	P	P	P	.	.	.	.	.	P	P	P	P	P	P	P	23	
25	KALE ROHINI KAILAS	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	.	.	P	26	
26	KAPRE KOMAL SUBHASH	P	P	P	P	P	P	P	P	P	P	P	.	.	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	25	

A. R. B. Garud Art's, Commerce and Science College, Shendurni.
Tal- Jamner, Dist- Jalgaon. Pin- 424204
Test for Chemistry

A. R. B. Garud Art's, Commerce and Science College, Shendurni.
Tal- Jamner, Dist- Jalgaon. Pin- 424204
Test for Chemistry

27	KASTURE MEGHA MANOJ																																		
28	KHAIRNAR KALPESH SANJAY	P	P	P	P	P	.	.	.	P	P	.	.	.	.	P	P	P	P	P	.	P	P	P	-	P	P	P	.	P	P	P	21	Khairnar	
29	KSHIRSAGAR VISHNU ASHOK	P	P	-	-	-	P	-	P	P	P	P	-	-	-	-	P	P	P	P	P	P	P	P	P	-	P	-	P	P	P	P	21		
30	KUMAVAT NITEEN HARI	P	P	P	P	P	.	.	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P	.	.	.	P	P	P	P	P	.	23		
31	LINGAYAT KISHOR RAJU	P	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P	.	.	.	P	P	P	P	.	.	P	P	P	P	.	.	21		
32	MAHAJAN NILESH SAMADHAN	P	P	P	P	P	P	P	P	P	P	.	.	.	.	.	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P	24		
33	MALI SAGAR RAMESH	P	P	P	P	P	.	.	.	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	P	.	.	.	P	P	23		
34	MALI SWATI VASANTA	P	P	P	.	.	P	P	P	.	.	P	P	P	.	.	P	P	P	.	.	P	P	P	.	.	P	P	P	P	P	P	21		
35	NAIK AASHISH PRABHU	P	P	P	.	.	P	P	P	.	.	P	P	P	.	.	P	P	P	.	.	P	P	P	.	.	P	P	P	P	P	P	21		
36	NIKAM RAVINDRA BHIMRAO	P	P	P	P	P	.	.	.	P	P	.	.	.	.	P	P	P	P	P	.	P	P	P	.	.	P	P	P	P	P	P	21		
37	NILPAGARE SACHIN DNYANESHWAR	P	P	P	P	P	.	.	.	P	P	P	P	.	.	P	P	P	P	P	.	P	P	P	.	P	P	P	.	P	P	P	23		
38	PARDESHI ISHWAR JAGDISH	P	P	P	P	P	.	P	P	P	P	P	-	.	.	.	.	P	P	P	P	P	.	.	P	.	P	P	P	P	P	P	21		
39	PARDESHI JAYAVANT DILIP	P	P	P	P	P	P	.	.	P	P	P	.	.	.	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P			
40	PATEL MANASI TEJAS	P	P	P	P	P	P	.	.	.	.	.	.	.	P	P	P	P	P	P	P	P	.	P	P	P	P	P	.	P	P				
41	PATIL ARATI PANKAJ	P	P	P	.	P	P	.	P	P	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	P	P				
42	PATIL BHAGYASHRI RAJENDRA	P	P	P	P																														
43	PATIL DIPAK RAJU																																		
44	PATIL DIVYA CHANDRAKANT																																		
45	PATIL KUNAL BHAGWAN																																		
46	PATIL MANOJ BHIKA																																		
47	PATIL MAYUR VISHWANATH																																		
48	PATIL PALLAVI SANJAY																																		
49	PATIL SANJANA BANDU																																		
50	PATIL SWATI EKNATH																																		
51	PATIL VAIBHAV ANANDA																																		
52	PAWAR MADHURI SATISH																																		
53	PAWAR RAHUL RAMESH																																		
54	PHUSE SHUBHANGI SANJAY																																		
55	PINJARI SHAKIL KADIR																																		
56	RATHOD HIMMATSING JAGANNATH																																		
57	SABANE KALYANI SHIVRAM																																		
58	SAIYYAD SIMIN AKHTAR NASIR ALI																																		
59	SAPKAL SHUBHAM MURLIDHAR																																		
60	SHAIKH ARBAAJ YUNUS																																		
61	SONET RAJ RUMSING																																		
62	TADAVI SADIK AYYUB																																		

63	USHIR VAIBHAV BHAGWAN	.	.	.	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	29	
64	VIKHANKAR VIJAY SUDHAKAR	P	P	P	.	.	.	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	24	
65	WAGH SHUBHAM VILAS	.	.	.	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		
66	AAGE HARSHAL YUVARAJ	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		
67	BANKAR AKASH VIKAS	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P		
68	BANKAR MAMATA VITTHAL	P	P	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		
69	DEKE BHUSHAN VASUDEV	P	P	P	P	P	P	.	P	P	P	.	.	P	P	P	P	P	P	P	-	.	.	P	P	P	P	P	P	P	P	P	P	25	
70	DHANGAR RUPALI KIRAN	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P		
71	GHOPE KANCHAN NAMDEO	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	.	.	P	P	P	P	P		
72	JADHAV MANOJ MOTILAL	P	P	P	P	P	P	P	.	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		
73	KHAKARE SNEHAL VILASH	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P		
74	MALI AKASH SUDHAKAR	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	.	.	P	P	P	P	P	P		
75	MALI PRAMOD KAILASH	P	P	P	P	.	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		
76	PATIL NILESH HIRALAL	P	P	P	P	P	P	P	P	P	P	P	.	.	.	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P		
77	SHRAWANE ASHWINI SHAMRAO	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	.	.	.	P	P	P	P		
78	SHRUNGARE PRAGATI BHAVADU	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		
79	BARI TEJAS NANDAKISHOR	P	P	P	P	P	.	.	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		
80	BAVSKAR SACHIN GUNVANT	P	P	P	P	P	P	P	P	P	P	P	.	.	.	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P		
81	BHOI MUKESH SANTOSH	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	.	.	.	.	P	P	P	P		
82	CHAVAN ABHISHEK PRAKASH	P	P	P	P	P	P	-	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		
83	MALI HEMANT DNYANDEV	P	P	-	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P		
84	PARDESHI GAJANAN SUVARNASING	P	P	P	P	P	P	P	.	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	26	
85	PATHAN SHOHEB MUKTAR	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	.	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	25	
86	PAWAR MANOJ RAMDAS	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	.	.	P	P	P	P	P	24	
87	SHAIKH MOHAMMAD TOHID MUSHTAK	P	P	P	P	P	P	P	P	P	.	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	26	
88	TELI DEVANAND VITTHAL	P	P	P	P	.	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	26	
89	THORAT DEEPALI VIJAY	P	P	P	P	P	P	P	P	P	P	P	P	P	.	.	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	P	26	

Subject Teacher

Principal

Appr... Art's
Comm... College
Shendurni-... Jamner

A. R. B. Garud Art's, Commerce and Science College, Shendurni.
Tal- Jamner, Dist- Jalgaon. Pin- 424204

Test for Chemistry

30

Select an appropriate option for each of the following

1) The ideal gas equation for 1 mole of gas is----
A] $PV = nRT$ B] $\frac{PV}{nT} = R$ C] $PV = RT$ D] None

2) Which of the following mostly account for the mass of an atom?
A] Neutrons B] Neutrons & Protons C] Electron & Proton D] Electron & Neutron

3) Saturated hydrocarbon are known as
A] Alkanes B] Alkenes C] Alkynes D] Alkaloids

4) The Van der Waal equation of gas is---
A] $PV = nRT$ B] $\left(P+\frac{a}{V^2}\right)(V-nb)= nRT$ C] $\left(P+\frac{a}{V^2}\right)(V-nb^2)= nRT$ D] None

5) Which metal was used by Rutheford in his α- scattering experiment
A] Gold B] Platinum C] Silver D] Lead

6) Identify the smallest alkane
A] Cyclomethane B] Methane C] Cyclopropane D] Propane

7) Van der Waal's equation explains behavior of –
A] Ideal gas B] Real gas C] Nobel gas D] liquefaction of gas

8) An element has an electronic configuration of (2,8,7) it's valency is –
A] 2 B] 7 C] 3 D] 1

9) Dienes are the name given to compounds with –
A] Exactly a double bond B] Exactly a triple bond
C] Exactly two double bond D] More than two double bond

10) Critical temperature of gas is the ___
A] Temp. above which a liquid can not exist B] Temp. below which a liquid can not exist
C] Minimum temp. at which a gas can be liquefied D] Maximum temp. at which a gas can be liquefied

11) During a chemical reaction, atomic number ---
A] Changes B] Remains same
C] Changes and then is restored D] Changes alternately

12) The substituent in the chain is named by replacing the 'ane' in the alkane by ---
A] ene B] ic
C] one D] yl

13) The gas will show ideal beahaviour at –
A] Low temp. and low pressure B] Low temp. and high pressure
C] High temp. and low pressure D] High temp. and high pressure

14) Which isotope of H contains only one proton and no neutron in its nucleus?
A] Protium B] Deuterium
C] Tritium D] None of the above

Principal
Approached to A. R. B. Garud Art's
Commerce & Science College
Shendurni-424204, Tal.Jamner

15) Which of the following is used for artificial ripening of fruit-
A] Ethylene B] Acetylene C] Phenol D] Methanol

16) The gas which shows max. deviation from ideal behavior is ----
A] CO_2 B] O_2 C] He D] N_2

17) The electronic configuration of an atom having atomic number 20 is –
A] (2,8,8,2) B] (2,8,10) C] (2,6,8,4) D] (2,4,8,6)

18) General formula for amine is –
A] $R\text{-}CH_2$ B] R_2 - CH C] R – NH_2 D] R - COOH

19) 'When a gas under high pressure is allow to expand adiabatically through a fine hole into a re low pressure is accompanied by cooling' , the effect is known as-
A] Joule-Thomson effect B] Linde's effect
C] Temperature effect D] None

20) An atom differs from its ion in
A] Nuclear charge B] Mass number C] Number of electrons D] Number of Neut

21) What is the general formula of alkenes-
A] C_nH_{2n+2} B] C_nH_{2n} C] C_nH_{2n-2} D] None

22) The critical temperature for CO_2 gas is ---- ^oC
A] 32 ^oC B] 13.1 ^oC C] 48 ^oC D] 31.1 ^oC

23) Which of the following is not fundamental particle-
A] Proton B] Neutron C] Alpha particle D] Electron

24) Which of the following shows molecular formula of methane-
A] CH_4 B] C_2H_6 C] C_3H_8 D] C_4H_{10}

25) The compressibility factor at the same temperature and pressure is given by—
A] $Z= \frac{PV}{nRT}$ B] $Z= \frac{V_{real}}{V_{ideal}}$ C] Z= nPVR D] None

26) Atom which have same mass number but different atomic number are called ---
A] Isotopes B] Isomer C] Isotones D] Isobars

27) An alkane with Molecular Formula C_7H_{16} is ---
A] Butane B] Pentane C] Hexane D] Heptane

28) If Z > 1 the gas is ---
A] More compressible B] Not compressible C] Less compressible D] None

29) How many electron does the outermost shell of Neon have?
A] 4 B] 8 C] 2 D] 16

30) Alcohols (CnH2n+1OH) are formed by attachment of
A] –OH group to Hydrocarbon chain B] H-X group to Hydrocarbon chain
C] O to Hydrocarbon chain D] H to Hydrocarbon chain

BEST OF LUCK

A. R. B. Garud Art's, Commerce and Science College, Shendurni.

Tal- Jamner, Dist- Jalgaon. Pin- 424204

ANSWER KEY

[Test For Chemistry]

Question No.	Responce	Question No.	Responce
1	C	16	A
2	B	17	A
3	A	18	C
4	B	19	A
5	A	20	C
6	B	21	B
7	B	22	D
8	D	23	C
9	C	24	A
10	D	25	B
11	B	26	D
12	D	27	D
13	C	28	C
14	A	29	B
15	B	30	A

PAPER SETTER

1] MR. S. E. DRAKSHE

2] MR. R. S. GARUD

Principal
Appasaheb R. B. Garud Art's
Commerce & Science College
Shendurni-424204,Tal.Jamner

ANSWER SHEET

CODE: CH-2017 A

INSTRUCTIONS: • Please read the instruction carefully before writing • Write only the Roll number, Question paper code and Answer • Darken ONLY ONE CIRCLE with BLACK PEN to mark your answer as shown in the example. If you darken more than one circle, your answer is treated as wrong. • Do not make any stray marks on the answer sheet. • Rough work must not be done on the answer sheet. • Do not fold the Answer Sheet

WRONG	RIGHT
⊗ ⊘ ◐ ◐	⊙ ⊙ ● ◐

DETAILS OF STUDENT –

NAME: PATIL KUNAL BHAGWAN

CLASS: FYBSC

SUBJECT: CHEMISTRY

DATE OF EXAM: 28 08 17

CENTRE OF EXAM: SHENDURNI

CODE: 170033

TOTAL: 14

Invigilator's Signature

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Principal
Appasaheb Birnale Arts
Commerce & Science College
Shendurni-416206, Tal. Jamner

ANSWER SHEET

Code : CH-2017A

INSTRUCTIONS : • Kindly read the instruction carefully before writing • Please ONLY ONE CIRCLE with PENCIL for [illegible] Roll number, Question paper code and [illegible] • Darken the circle your answer [illegible] answer or [illegible] in the example. If you darken more than the circle your answer [illegible] treated as [illegible] • Do not make any stray marks on the answer sheet • Rough work must not be done on answer sheet • Do not fold the Answer Sheet

WRONG	RIGHT
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DETAILS OF STUDENT –

NAME PATIL MANOJ BHIKA

CLASS FYBSC

SUBJECT CHEMISTRY

DATE OF EXAM 28 08 17

CENTRE OF EXAM SHENDURNI

CODE 170033

TOTAL 16

Invigilator's Signature

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Principal
Appasaheb [illegible] Arts
Commerce [illegible] College
Shendurni [illegible]

Appasaheb Ragunathrao Bhaurao Garud Arts, Commerce and Science College, Shendurni.

Pre/Post Test for bridge course

Slow And Advance learner

Subject: **Academic Year :2018-2019** **Class: F. Y. B. Sc.** **Month:**

Roll No	Student Name	Test- 1	Test- 2
1	AUTI AKSHAY SHYAM	19	
2	BADGUJAR MAHENDRA MOHAN	08	20
3	BANKAR JITENDRA DNYANESHWAR	AB	
4	BARI BHUSHAN SUNIL	14	
5	BARI DEVENDRA DIGAMBAR	19	
6	BARI KAJAL RAJENDRA	AB	
7	BARI PRAJAKTA NANDLAL	15	
8	BARI PUNAM GOPAL	18	
9	BARI PUSHPRAJ SHIVRAM	AB	
10	BARI RUPALI DNYANDEO	AB	
11	BARI VAIBHAV SURESH	10	23
12	BERAD NAYAN SUNIL	11	
13	BHAMARE SAURAV ARUN	11	
14	BHOI HEMANT VIJAY	27	
15	BHOSANDE KOMAL DNYANESHWAR	AB	
16	BODADE ARATI PANDURANG	17	
17	CHAUDHARI GAYATRI SANJAY	16	
18	CHAUDHARI KRUNAL SHAM	12	
19	CHAUDHARI SWATI SANJAY	20	
20	CHAVAN SHIVANI PRAVIN	14	
21	CHAVAN SHUBHANGI PRAVIN	28	
22	DEVARE PUNAM KISAN	11	
23	DHANGAR NIRMALA SHAMRAO	16	
24	GAYAKWAD PRATIKSHA ARJUN	18	
25	GHOLAP ASHWINI VINOD	14	
26	GHONGADE ROHINI DILIP	17	
27	GUJAR SANJANA YUVRAJ	29	
28	KALE GOPAL BHAGWAN	18	
29	KAPASE SAURABH VIJAY	15	
30	KHALSE BHUSHAN MADHUKAR	08	25
31	KOLI GAJANAN GANGARAM	AB	
32	KOLI TANVI MACHHINDRA	22	

Principal
Appasaheb Ragunathrao Garud Art's Commerce & Science College
Shendurni

33	KULKARNI RAJESH DIGAMBAR	15	
34	KUMAVAT PALLAVI SUBHASH	27	
35	KUMAVAT POONAM VILAS	18	
36	MALI RITESH VITTHAL	AB	
37	MALI TUSHAR SAMADHAN	08	27
38	PATIL AKSHAY SHIVAJI	13	
39	PATIL ALPESH SANJAY	20	
40	PATIL BHAVESH SAMBHAJI	AB	
41	PATIL NISHA PREMSING	24	
42	PATIL ROHINI KISHOR	16	
43	PATIL RUSHIKESH DHARMARAJ	14	
44	PATIL SHAILESH SHENFADU	08	22
45	PAWAR JYOTSNA VIJAY	13	
46	RATHOD AJAY BADRINARAYAN	11	
47	RAUT MINAKSHI SAMADHAN	16	
48	RONDHALE JYOTI HARI	17	
49	SAPKAL KUNAL RAJENDRA	12	
50	SAYYAD MUBASHSHIRA FIRDOS ALTAF ALI	AB	
51	SHELKE AMOL NANA	17	
52	SHELKE SAURABH GAJANAN	15	
53	SHRISHRIMAL AKANKSHA SHAMLAL	AB	
54	SURWADE KIRAN KADUBA	11	
55	THATE NITA ARJUN	22	
56	TUPE KALPESH SHRAWAN	16	
57	AMBIKAR AMOL RAJENDRA	AB	
58	BHADANE SNEHAL ANIL	14	
59	GITE ASHISH ASHOK	19	
60	GITE HARSHAL SANJAY	16	
61	PATIL AKSHAY CHANDRAKANT	AB	

Appasaheb Dharmaraj Arts, Commerce & Science College
Shendurni-424304 Jalgaon

Appasaheb Ragunathrao Bhaurao Garud Arts, Commerce and Science College, Shendurni.

Pre/Post Test for bridge course
Slow And Advance learner

Subject: **Academic Year :2018-2019** **Class: F. Y. B. Sc.** **Month:** September (Including Sunday)

Roll No	Student Name	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	Total	Sign Of Student
1	AUTI AKSHAY SHYAM																																	
2	BADGUJAR MAHENDRA MOHAN	P	P	P	P	P	-	-	P	P	P	P	-	P	P	P	P	P	P	P	P	-	P	P	-	-	P	P	-	-	P		21	
3	BANKAR JITENDRA DNYANESHWAR	-	P	P	-	-	P	P	-	P	P	P	P	P	-	P	P	P	P	-	P	P	P	P	-	P	P	P	-	-	-		20	
4	BARI BHUSHAN SUNIL	P	-	-	-	P	P	P	P	P	-	P	P	P	-	P	P	P	P	-	P	-	-	-	-	P	P	-	-	P	P		18	
5	BARI DEVENDRA DIGAMBAR	-	P	P	P	P	P	P	P	P	P	-	-	-	P	P	-	P	P	P	P	-	P	P	P	-	-	P	P	P	P		22	
6	BARI KAJAL RAJENDRA	P	P	P	P	P	-	-	P	P	P	-	-	P	P	-	P	P	P	P	P	P	P	P	-	P	P	P	P	P	P		24	
7	BARI PRAJAKTA NANDLAL	P	P	P	-	P	P	-	-	-	P	P	P	P	P	P	P	P	-	P	P	P	-	P	P	P	P	P	P	P	P		23	
8	BARI PUNAM GOPAL	P	P	P	P	P	P	P	P	-	-	P	P	P	P	P	P	P	P	P	-	P	P	P	P	P	P	P	P	-	P		25	
9	BARI PUSHPRAJ SHIVRAM	P	P	P	-	P	-	P	-	P	-	P	P	P	P	P	P	P	P	P	P	-	P	P	-	-	P	P	P	P	P		23	
10	BARI RUPALI DNYANDEO	-	-	P	P	P	P	P	P	-	P	P	P	P	P	P	P	P	P	P	P	P	-	-	P	P	P	P	-	-	P		25	
11	BARI VAIBHAV SURESH	P	P	P	P	P	P	P	P	-	-	-	P	P	P	P	P	P	P	P	-	-	P	P	P	P	P	-	P	P	P		24	
12	BERAD NAYAN SUNIL	P	P	P	P	P	-	-	P	P	P	P	P	P	P	P	-	P	P	P	P	P	P	P	P	P	P	P	P	-	-		22	
13	BHAMARE SAURAV ARUN	P	P	P	P	P	P	P	-	-	P	P	P	P	P	P	-	-	P	P	P	-	-	P	P	P	P	P	P	P	P		25	
14	BHOI HEMANT VIJAY	-	-	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P	-	-	P	P	P	P	P	P	P	P	P	P	P		25	
15	BHOSANDE KOMAL DNYANESHWAR	P	P	P	P	P	P	-	.	P	P	P	P	P	P	.	P	P	P	P	.	.	P	P	P	P	P	-	P	P	P		24	
16	BODADE ARATI PANDURANG	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	-	-	P	P	P	P		28	
17	CHAUDHARI GAYATRI SANJAY	P	P	P	P	P	P	P	P	P	P	P	P	P	-	.	P	P	P	P	P	P	-	-	-	P	P	P	P	P	P		25	
18	CHAUDHARI KRUNAL SHAM	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	P	P	P	.	P	P	P	P	-	P	P	P	P	P		26	
19	CHAUDHARI SWATI SANJAY	P	P	P	P	P	-	-	P	P	P	P	P	P	P	P	-	-	P	P	P	P	P	P	-	-	P	P	P	P	-		28	
20	CHAVAN SHIVANI PRAVIN	P	P	P	P	P	P	P	P	P	P	P	P	P	-	.	P	P	P	P	-	-	P	P	P	P	P	-	P	P	P		25	
21	CHAVAN SHUBHANGI PRAVIN	P	P	P	P	-	.	P	P	P	P	P	P	P	.	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P	-		24	
22	DEVARE PUNAM KISAN	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	-	P	P	P	P	-	.	P	P	P	P	P	P	P	-		27	
23	DHANGAR NIRMALA SHAMRAO	-	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P	.	P	P	P	-	P	P	P	P	P	P	P	-	-		24	
24	GAYAKWAD PRATIKSHA ARJUN	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	-	.	P	P	-	-	P	P	P		26	
25	GHOLAP ASHWINI VINOD	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	.	.	P	P	P	P	P	P	P	-	.	P	P	P	P		24	
26	GHONGADE ROHINI DILIP	P	P	P	P	P	P	P	P	P	P	P	-	-	P	P	P	P	P	P	P	P	P	P	P	.	.	P	P	P	P		26	
27	GUJAR SANJANA YUVRAJ	P	P	P	P	P	P	P	P	P	P	P	P	P	-	-	P	P	P	P	P	P	P	P	P	-	-	.	.	.	.		23	
28	KALE GOPAL BHAGWAN																																	
29	KAPASE SAURABH VIJAY	P	P	P	P	P	P	P	P	P	P	-	.	P	P	.	.	P	P	P	P	P	.	P	P	P	P	P	P	P	P		25	
30	KHALSE BHUSHAN MADHUKAR	P	P	P	P	P	P	.	P	P	P	P	-	P	P	P	.	.	P	P	P	P	P	P	P	P	P	P	P	P	P		26	

31	KOLI GAJANAN GANGARAM																																
32	KOLI TANVI MACHHINDRA	P	P	P	-	-	P	P	P	P	-	-	-	P	P	P	P	P	P	P	P	P	P	P	P	-	-	P	P	P	P	P	23
33	KULKARNI RAJESH DIGAMBAR																																
34	KUMAVAT PALLAVI SUBHASH	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	30
35	KUMAVAT POONAM VILAS	-	-	P	P	P	P	P	P	P	P	-	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	-	25
36	MALI RITESH VITTHAL	-	-	-	P	P	P	P	P	P	P	P	P	P	-	-	P	P	P	P	P	-	-	P	P	P	P	P	P	P	P	P	23
37	MALI TUSHAR SAMADHAN	P	P	P	P	P	P	P	P	P	P	P	P	P	-	-	P	-	-	P	P	P	P	-	-	P	P	P	P	P	P	-	23
38	PATIL AKSHAY SHIVAJI	P	P	P	P	P	P	-	P	P	P	P	-	-	P	P	P	P	P	P	P	P	P	P	P	-	-	P	P	P	P	P	24
39	PATIL ALPESH SANJAY	P	P	P	P	-	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	-	-	P	P	P	-	P	25
40	PATIL BHAVESH SAMBHAJI	P	P	P	P	P	P	P	P	-	P	P	P	P	P	P	P	P	P	P	-	-	P	P	P	P	-	-	P	P	P	-	24
41	PATIL NISHA PREMSING	-	P	P	P	P	P	P	P	P	P	P	-	P	P	P	P	P	P	P	P	-	-	P	P	-	P	P	P	P	-	P	25
42	PATIL ROHINI KISHOR	P	P	P	P	P	P	P	P	P	-	P	P	P	P	P	-	P	P	P	P	P	P	P	-	P	P	P	-	P	P		23
43	PATIL RUSHIKESH DHARMARAJ	P	P	P	P	-	P	P	P	-	P	P	P	P	P	P	P	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P	27
44	PATIL SHAILESH SHENFADU	P	P	P	P	-	P	P	P	P	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P	-	P	P	P	P	P	P	27
45	PAWAR JYOTSNA VIJAY	P	P	P	P	P	P	P	P	P	P	P	-	-	-	-	-	P	P	P	P	P	P	P	P	P	P	P	P	-	-	P	22
46	RATHOD AJAY BADRINARAYAN	P	P	P	-	-	P	P	P	P	P	P	-	P	P	P	P	P	P	P	P	P	P	P	P	-	-	-	-	P	P	P	23
47	RAUT MINAKSHI SAMADHAN	P	P	P	P	P	P	P	P	P	P	P	-	-	P	P	P	P	P	P	P	P	P	P	-	-	P	P	P	P	-	-	24
48	RONDHALE JYOTI HARI	-	-	-	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	26
49	SAPKAL KUNAL RAJENDRA																																
50	SAYYAD MUBASHSHIRA FIRDOS ALTAF ALI																																
51	SHELKE AMOL NANA	P	P	P	P	P	P	P	-	-	P	P	P	P	P	P	P	P	P	P	P	-	-	-	P	P	P	-	-	P	P	P	23
52	SHELKE SAURABH GAJANAN	P	P	P	P	P	P	P	P	-	-	-	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	25
53	SHRISHRIMAL AKANKSHA SHAMLAL	P	P	P	P	P	-	-	-	P	P	P	P	P	P	P	P	-	-	P	P	P	P	P	P	P	P	P	P	-	-	P	23
54	SURWADE KIRAN KADUBA	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	-	-	-	P	P	P	P	-	-		25
55	THATE NITA ARJUN	P	P	P	P	P	-	-	-	P	P	P	P	P	P	P	P	P	P	-	-	P	P	P	P	P	P	P	P	-	-	P	25
56	TUPE KALPESH SHRAWAN	P	P	P	-	-	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	-	-	P	P	P	-	-	P	P	P	-	23
57	AMBIKAR AMOL RAJENDRA																																
58	BHADANE SNEHAL ANIL	-	-	-	P	P	P	P	-	-	-	P	P	P	P	P	P	P	P	-	-	P	P	P	-	-	P	P	P	P	P	-	21
59	GITE ASHISH ASHOK																																
60	GITE HARSHAL SANJAY																																
61	PATIL AKSHAY CHANDRAKANT	P	P	P	-	-	-	P	P	P	P	P	P	P	P	P	P	P	P	P	-	-	P	P	P	P	-	-	P	P	P	P	23

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Test for Chemistry

elect an appropriate option for each of the following **30**

) The ideal gas equation for 1 mole of gas is----

A] PV = nRT B] $\frac{PV}{nT} = R$ C] PV = RT D] None

) Which of the following mostly account for the mass of an atom?

A] Neutrons B] Neutrons & Protons C] Electron & Proton D] Electron & Neutron

) Saturated hydrocarbon are known as

A] Alkanes B] Alkenes C] Alkynes D] Alkaloids

4) The Van der Waal equation of gas is---

A] PV = nRT B] $\left(P+\frac{a}{V^2}\right)(V-nb)= nRT$ C] $\left(P+\frac{a}{V^2}\right)(V-nb^2)= nRT$ D] None

5) Which metal was used by Rutheford in his α- scattering experiment

A] Gold B] Platinum C] Silver D] Lead

6) Identify the smallest alkane

A] Cyclomethane B] Methane C] Cyclopropane D] Propane

7) Van der Waal's equation explains behavior of –

A] Ideal gas B] Real gas C] Nobel gas D] liquefaction of gas

8) An element has an electronic configuration of (2,8,7) it's valency is –

A] 2 B] 7 C] 3 D] 1

9) Dienes are the name given to compounds with –

A] Exactly a double bond B] Exactly a triple bond
C] Exactly two double bond D] More than two double bond

10) Critical temperature of gas is the ___

A] Temp. above which a liquid can not exist B] Temp. below which a liquid can not exist
C] Minimum temp. at which a gas can be liquefied D] Maximum temp. at which a gas can be liquefied

11) During a chemical reaction, atomic number ---

A] Changes B] Remains same
C] Changes and then is restored D] Changes alternately

12) The substituent in the chain is named by replacing the 'ane' in the alkane by ---

A] ene B] ic
C] one D] yl

13) The gas will show ideal beahaviour at –

A] Low temp. and low pressure B] Low temp. and high pressure
C] High temp. and low pressure D] High temp. and high pressure

14) Which isotope of H contains only one proton and no neutron in its nucleus?

A] Protium B] Deuterium
C] Tritium D] None of the above

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15) Which of the following is used for artificial ripening of fruit-
A] Ethylene B] Acetylene C] Phenol D] Methanol

16) The gas which shows max. deviation from ideal behavior is ----
A] CO_2 B] O_2 C] He D] N_2

17) The electronic configuration of an atom having atomic number 20 is –
A] (2,8,8,2) B] (2,8,10) C] (2,6,8,4) D] (2,4,8,6)

18) General formula for amine is –
A] $R\text{-}CH_2$ B] R_2 - CH C] R – NH_2 D] R - COOH

19) 'When a gas under high pressure is allow to expand adiabatically through a fine hole into a region of low pressure is accompanied by cooling' , the effect is known as-
A] Joule-Thomson effect B] Linde's effect
C] Temperature effect D] None

20) An atom differs from its ion in
A] Nuclear charge B] Mass number C] Number of electrons D] Number of Neutrons

21) What is the general formula of alkenes-
A] C_nH_{2n+2} B] C_nH_{2n} C] C_nH_{2n-2} D] None

22) The critical temperature for CO_2 gas is ---- ^{O}C
A] 32 ^{O}C B] 13.1 ^{O}C C] 48 ^{O}C D] 31.1 ^{O}C

23) Which of the following is not fundamental particle-
A] Proton B] Neutron C] Alpha particle D] Electron

24) Which of the following shows molecular formula of methane-
A] CH_4 B] C_2H_6 C] C_3H_8 D] C_4H_{10}

25) The compressibility factor at the same temperature and pressure is given by—
A] $Z= \frac{PV}{nRT}$ B] $Z= \frac{V_{real}}{V_{ideal}}$ C] Z= nPVR D] None

26) Atom which have same mass number but different atomic number are called ---
A] Isotopes B] Isomer C] Isotones D] Isobars

27) An alkane with Molecular Formula C_7H_{16} is ---
A] Butane B] Pentane C] Hexane D] Heptane

28) If Z > 1 the gas is ---
A] More compressible B] Not compressible C] Less compressible D] None

29) How many electron does the outermost shell of Neon have?
A] 4 B] 8 C] 2 D] 16

30) Alcohols (CnH2n+1OH) are formed by attachment of
A] –OH group to Hydrocarbon chain B] H-X group to Hydrocarbon chain
C] O to Hydrocarbon chain D] H to Hydrocarbon chain

BEST OF LUCK

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ANSWER KEY

[Test For Chemistry]

Question No.	Responce	Question No.	Responce
1	C	16	A
2	B	17	A
3	A	18	C
4	B	19	A
5	A	20	C
6	B	21	B
7	B	22	D
8	D	23	C
9	C	24	A
10	D	25	B
11	B	26	D
12	D	27	D
13	C	28	C
14	A	29	B
15	B	30	A

PAPER SETTER
1] MR. S. E. DRAKSHE

2] MR. R. S. GARUD

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ASS FYBSC

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Invigilator's Signature

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AILS OF STUDENT –

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NTRE OF EXAM 21254 CODE 170033

TOTAL 22

Invigilator's Signature

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DETAILS OF STUDENT –

NAME: KALE GOPAL B

CLASS: FYBSC

SUBJECT: CHEMISTRY

DATE OF EXAM: 20 08 18

CENTRE OF EXAM: SHENDURNI

CODE: 170033

TOTAL: 18

Invigilator's Signature

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92	O	O	O	O
93	O	O	O	O
94	O	O	O	O
95	O	O	O	O
96	O	O	O	O
97	O	O	O	O
98	O	O	O	O
99	O	O	O	O
100	O	O	O	O

Appasaheb Dharmadhikari's Commerce & Science College Shendurni-424304, Tal. Jamner

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ANSWER KEY

[Test For Chemistry]

Question No.	Responce	Question No.	Responce
1	C	16	A
2	B	17	A
3	A	18	C
4	B	19	A
5	A	20	C
6	B	21	B
7	B	22	D
8	D	23	C
9	C	24	A
10	D	25	B
11	B	26	D
12	D	27	D
13	C	28	C
14	A	29	B
15	B	30	A

Principal
Appasaheb R. B. Garud Art's
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PAPER SETTER
1] MR. S. E. DRAKSHE

2] MR. R. S. GARUD